

(1*R*,2*R*)-1,3-Bis(*tert*-butyldimethylsilyloxy)-2-(4-hydroxybutanamido)-1-(4-nitrophenyl)-propane**Dorte Johansson, Frank B. Larsen, Andrew D. Bond and Poul Nielsen***

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Key indicators

Single-crystal X-ray study

 $T = 180\text{ K}$ Mean $\sigma(\text{C}-\text{C}) = 0.012\text{ \AA}$

Disorder in main residue

 R factor = 0.056 wR factor = 0.148

Data-to-parameter ratio = 6.3

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.The title compound, $\text{C}_{25}\text{H}_{46}\text{N}_2\text{O}_6\text{Si}_2$, is a new derivative of chloramphenicol. At 180 K, the structure has $Z' = 2$ in space group $P2_1$.

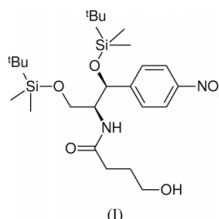
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Comment

Chloramphenicol is produced naturally by the soil bacterium *Streptomyces venezuelae* and is a widely used antibiotic in the treatment of eye infections (Long & Porse, 2003); it blocks the peptidyl transferase activity by hampering the binding of transfer RNA to the A-site in the ribosome (Schlünzen *et al.*, 2001). The synthesis of biologically active analogues of chloramphenicol has attracted chemists since preparation of the first examples in 1951 (Huebner & Scholtz, 1951). The title compound, (I), was synthesized as part of our effort to produce new chloramphenicol analogues. It is made in two steps from (1*R*,2*R*)-2-amino-1-(4-nitrophenyl)-1,3-propanediol in 66% yield. This is the first structural report of a chloroamphenicol derivative bearing a hydroxybutanoyl substituent on the amine N atom.



In space group $P2_1$, there are two independent molecules in the asymmetric unit. The conformations of these two molecules are largely comparable, but differ in the orientation of the hydroxybutanoyl substituents. In molecule *A*, the hydroxyl group O6*A* is brought into a position effectively *cis* to the carbonyl oxygen O5*A*. In molecule *B*, O6*B* lies effectively *trans* to O5*B*. The *tert*-butyldimethylsilane (TBDMS) groups containing Si2*A/B* also differ in their orientation: the torsion angles $\text{C5A}-\text{O3A}-\text{Si2A}-\text{C23A}$ and $\text{C5B}-\text{O3B}-\text{Si2B}-\text{C23B}$ are $174.7(6)$ and $107.2(6)^\circ$, respectively. Viewed in projection on to the plane of the benzene ring, the $\text{C4A}-\text{C5A}$ bond in molecule *A* lies approximately parallel to one $\text{Si}-\text{CH}_3$ bond ($\text{Si2A}-\text{C18A}$) of the TBDMS group, while in molecule *B*, the $\text{C4B}-\text{C5B}$ bond lies approximately parallel to the bisector of the $\text{C18B}-\text{Si2B}-\text{C19B}$ angle. In both molecules, the *tert*-butyl group is staggered with respect to the $\text{Si}(\text{CH}_3)_2$ moiety.

Experimental

Compound (I) was synthesized by reaction of γ -butyrolactone with (1*R*,2*R*)-amino-1-(4-nitrophenyl)-1,3-bis(*tert*-butyldimethylsilyloxy)-

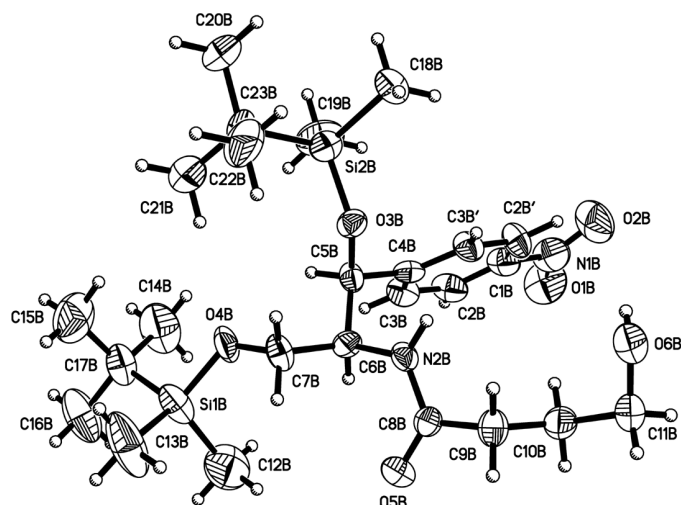


Figure 1

One symmetry-independent molecule in (I), showing displacement ellipsoids at the 50% probability level and H atoms drawn as spheres of arbitrary radius.

propane. Colourless crystals suitable for X-ray analysis were obtained by slow evaporation of a hexane solution. The melting point of (I) is 359–361 K.

Crystal data

$C_{25}H_{46}N_2O_6Si_2$	$D_x = 1.135 \text{ Mg m}^{-3}$
$M_r = 526.82$	Mo $K\alpha$ radiation
Monoclinic, $P2_1$	Cell parameters from 2489 reflections
$a = 14.9014 (12) \text{ \AA}$	$\theta = 2.4\text{--}18.4^\circ$
$b = 12.0241 (10) \text{ \AA}$	$\mu = 0.15 \text{ mm}^{-1}$
$c = 17.2103 (14) \text{ \AA}$	$T = 180 (2) \text{ K}$
$\beta = 91.288 (2)^\circ$	Needle, colourless
$V = 3082.9 (4) \text{ \AA}^3$	$0.50 \times 0.10 \times 0.10 \text{ mm}$
$Z = 4$	

Data collection

Bruker SMART 1000 CCD diffractometer	3985 independent reflections
Thin-slice ω scans	2691 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 2002)	$R_{\text{int}} = 0.076$
$T_{\text{min}} = 0.881$, $T_{\text{max}} = 0.985$	$\theta_{\text{max}} = 22.0^\circ$
12 674 measured reflections	$h = -15 \rightarrow 15$
	$k = -12 \rightarrow 7$
	$l = -18 \rightarrow 18$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0794P)^2 + 0.6178P]$
$R[F^2 > 2\sigma(F^2)] = 0.056$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.148$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.03$	$\Delta\rho_{\text{max}} = 0.33 \text{ e \AA}^{-3}$
3985 reflections	$\Delta\rho_{\text{min}} = -0.36 \text{ e \AA}^{-3}$
637 parameters	
H-atom parameters constrained	

The diffraction data were relatively weak and were truncated to $\theta = 22^\circ$ (0.95 Å resolution). As a result, the refined structure is of relatively low precision. The absolute configuration cannot be determined reliably and was assigned on the basis of the known chirality of

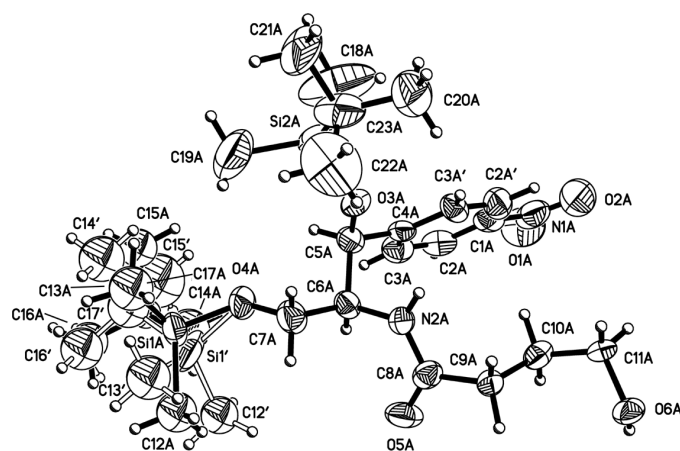


Figure 2

The second symmetry-independent molecule in (I), displaying disorder of one *tert*-butyltrimethylsilyl (TBDMS) group. Displacement ellipsoids are shown at the 50% probability level and H atoms are drawn as spheres of arbitrary radius.

the starting materials. Friedel opposites (1809 pairs) were merged in the final cycles of refinement. One *tert*-butyltrimethylsilyl (TBDMS) group is disordered and was modelled in two orientations. The geometries of these groups were restrained and the C atoms were refined with isotropic displacement parameters. H atoms bound to C atoms were positioned geometrically and allowed to ride during subsequent refinement, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ and distances of 0.95 Å for H atoms bound to benzene rings, 1.00 Å for H atoms bound to tertiary C atoms, 0.99 Å for H atoms of CH_2 groups, 0.88 Å for H atoms bound to N atoms, 0.84 Å for the H atom of the OH group and 0.98 Å for the H atoms of the methyl groups. In the last two cases, $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 2000); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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